

PII: S0040-4020(96)00883-6

Intramolecular Nitrone Cycloaddition: Stereoselective Synthesis of Piperidine Systems

Ugo Chiacchio, * Antonio Rescifina, * Francesco Casuscelli, * Anna Piperno, * Vincenzo Pisani, * and Roberto Romeo Casuscelli, * Anna Piperno, * Usani, * and Roberto Romeo Casuscelli, * Anna Piperno, * Usani, * and Roberto Romeo Casuscelli, * anna Piperno, * anna Pipern

^aDipartimento di Scienze Chimiche, Università, 95125 Catania, Italy

^bDipartimento Farmaco-Chimico, Università, 98168 Messina, Italy

°Dipartimento di Chimica, Università, 87036 Arcavacata di Rende, Italy

Abstract: A synthetic approach to isomerically functionalized piperidine systems has been designed by intramolecular nitrone cycloaddition, starting from β -enamidoaldehydes, and by subsequent reductive ring-opening of the obtained fused δ -lactams. Copyright © 1996 Published by Elsevier Science Ltd

Intramolecular 1,3-dipolar cycloadditions have recently received considerable synthetic and mechanistic interest as a convenient tool for the rapid construction of the complex carbon frameworks occurring in natural products and biological molecules.^{1,2}

A significant body of information regarding the regiochemistry and the stereochemistry of the process has been collected since the studies of Le Bel.³ The intramolecular 1,3-dipolar cycloaddition can be viewed as a competitive process between the bridged and fused modes of cycloaddition, controlled by a suitable interplay of factors such as alkene polarity, ring strain and other non-bonded interactions.⁴

Scheme 1

5-Hexen-1-imino-N-oxides 1 react with high regio- and stereoselectivity to give the fused 3-oxa-2-azabicyclo[3.3.0]octanes (cis) 2 as exclusive products, while 6-hepten-1-imino-N-oxides 3 gave predominantly fused products 8-oxa-7-azabicyclo[4.3.0]nonanes 4 (cis and trans fusion), together with the bridged 7-oxa-8-azabicyclo[4.2.1]nonanes 5 (Scheme 1). However, the obtained results suggest that steric effects at the reacting carbon termini may operate against fused-product regiochemistry.⁵

To gain a better understanding of the factors which control the balance between the two reaction courses, we have extended the process to a series of compounds in which an amido group has been inserted in the tether connecting the nitrone moiety to the dipolarophilic double bond. Recently we have reported that 5-hexen-3-*N*-methyl-4-oxo-1-imino-*N*-oxides 6 lead stereoselectively to *cis* fused γ-lactams 7 (Scheme 2).

Scheme 2

In this paper, we describe a novel general approach to functionalized fused δ -lactams 10 and 11 by intramolecular nitrone cycloaddition of 6-hepten-4-N-methyl-5-oxo-1-imino-N-oxides 8 and 6-hepten-3-N-methyl-4-oxo-1-imino-N-oxides 9 (Scheme 3).

Scheme 3

Our interest in these reactions is further promoted by the possibility of a selective functionalization of the obtained fused systems, by ring cleavage of isoxazolidine moiety, to give piperidine derivatives, which occur widely in a number of alkaloids.⁷

RESULTS AND DISCUSSION

The β -amido aldehydes 16 required for this study were synthesized as reported in scheme 4. The reaction of 3-(N-substituted-amino) propanol 12⁸ with α,β -unsaturated acyl chlorides 13 led to the corresponding amidoesters 14 which have been converted into the amido alcohols 15 by selective hydrolysis with K_2CO_3 in $H_2O/MeOH$. Swern-like oxidation of 15 afforded the corresponding aldehydes 16 which have been transformed by reaction with hydroxylamine derivatives to the corresponding 8-oxa-3,7-diazabicyclo[4.3.0]nonan-2-ones 18, via the not isolated nitrones 17, as exclusive products (Scheme 4).

OH
$$R_2$$
CH=CHCCI R_1 R_2 R_2 R_2 R_3 R_4 R_2 R_4 R_5 R_5

Scheme 4

The obtained derivatives were characterized on the basis of analytical and spectroscopic data. High resolution mass spectra showed the correct molecular ions. Ir absorptions of carbonyl groups are at 1685-1675 cm⁻¹ in accord with δ -lactams. The ¹H NMR spectra showed the H₉ proton in the range 3.89-5.27 δ , while H₁ and H₆ protons resonate at 2.80-3.50 δ . The methylene protons at C₄ showed different chemical shifts in the range 3.10-3.20 and 3.40-3.60 δ , while their resonances are coincident in the precursor amidoaldehydes 16; methylene protons at C₅ give rise to an indistinct multiplet centered at 1.80 δ .

The investigated 1,3-dipolar cycloaddition showed a complete regioselectivity: no bridged adducts have been detected in the crude reaction mixtures. The reactions have also been found to be stereospecific; intramolecular cycloadducts 18 were obtained stereochemically pure, with no evidence in the nmr spectra or tlc of the crude products of any diastereomers. The relative stereochemistry at C_1 - C_9 in the formed isoxazolidine ring is predetermined by the alkene geometry; furthermore, the ring junction between isoxazolidine and lactam 6-membered rings is always cis, as evidenced by NOE experiments. In fact, in compound 18a-c, irradiation of the upfield resonance corresponding to methylene protons at C_4 resulted in the observation of a signal enhancement for H_1 and H_6 ; similarly, when the resonance for H_6 was irradiated, a positive NOE effect was observed for H_1 and the ortho protons of the phenyl substituent at C_9 , in compounds 18 a,b, and for H_1 and the methyl group at C_9 , in compound 18c. On the contrary, irradiation of H_9 gives rise, in compounds 18a,b, to a NOE effect only for H_1 and the aromatic protons at C_9 .

These results are in agreement with the structure of stereoisomers 18a-c which show H_1 and H_6 in a syn relationship.

A different route has been exploited towards the synthesis of the analogous 8-oxa-3,9-diazabicyclo-[4.3.0]nonan-4-ones. The reaction pathway reported in scheme 5 starts from N-methylaminoacetaldehyde dimethylacetal 19 which was converted in the α -amido aldehydes 21 according to previously reported

procedure.⁶ Treatment of 21 with N-methyl hydroxylamine afforded a mixture of fused (cis and trans) compounds 23 and 24 (Scheme 5).

The structural assignments to obtained cycloadducts were readily made by MS, ¹H and ¹³C NMR analytical data (see Experimental).

The intramolecular cycloaddition process is always regiospecific leading to the exclusive formation of fused compounds as indicated by the presence of diagnostic ¹H NMR absorptions expected for isoxazolidine protons at C₇: no bridged products were detected in the crude reaction mixture either chromatographically or spectroscopically.

The stereochemical outcome of the intramolecular cycloaddition process appears of some interest: the nature of the ring fusion stereochemistry in isoxazolidines 23 (cis) and 24 (trans) is linked to the substitution pattern at the alkene carbon C_7 (R); thus, cis adducts result as major (cis/trans ratio = 70:30) or nearly exclusive products (cis/trans ratio = 99:1), when a methyl or a phenyl group respectively is incorporated at R; on the contrary, when R is an hydrogen atom, only trans adducts are obtained.

The proposed stereochemical features have been determined by analysis of ${}^{1}H$ NMR spectra and by NOE difference spectroscopy. In *cis* compounds **23b,c**, irradiation of the upfield H_{2} resonance gives rise to a positive NOE effect on H_{1} and H_{6} , so indicating a *cis* relationship between these protons. Furthermore, when H_{6} is irradiated, a comparable NOE for H_{1} and the methyl or phenyl substituent at C_{7} and, in a lesser degree, for the adjacent H_{7} is observed. Analogously, irradiation of H_{7} causes an enhancement of the resonance of the substituent at C_{7} and a lesser effect for the hydrogen atom at C_{6} .

On the contrary, in compound 24b, when H₇ is irradiated, a 3% NOE on H₁ is observed, so suggesting that these protons are topologically close in a *trans* ring fusion arrangement.

The stereochemical characteristics of *trans* compound 24a cannot be easily determined by NOE measurements and have been assigned by analysis of ${}^{1}H$ NMR data: in fact, the ${}^{1}H$ NMR spectrum shows a nearly coincident chemical shift for H_{1} and H_{6} (3.00 δ), besides a downfield resonance (4.20 δ) for one of the protons at C_{7} . These features can be considered diagnostic of a *trans* ring fusion in this kind of compounds as confirmed by the analysis of the ${}^{1}H$ NMR spectra of analogous compounds 23b and 24b; in fact, H_{1} and H_{6} resonate with different chemical shifts (2.47 δ) and (3.01 δ) in *cis* compound 23b and nearly at the same chemical shift in the *trans* isomer 24b; moreover H_{7} in 24b resonate at lower field (4.44 δ) with respect to the analogous proton in *cis* derivative 23b (3.69 δ).

Scheme 6

Functionalization of the obtained compounds 18, 23 and 24 have been performed by ring cleavage of isoxazolidine nucleus towards the formation of substituted piperidines. Thus, reduction of 18b, as model compound, with LiAlD₄, in anhydrous THF at reflux for 4 h, afforded the corresponding piperidine 25 in high yields (Scheme 6).

The obtained compounds gave satisfactory elemental analysis. The presence of NH and OH groups was indicated by IR absorptions at 3295 and 3420 cm⁻¹ respectively and by the presence of a broad singlet in the ^{1}H NMR spectrum integrating as two protons and exchangeable with $D_{2}O$.

In conclusion, a new synthetic approach to isomerically functionalized piperidine systems has been designed by intramolecular nitrone cycloaddition. The amino and alcoholic functionalities present in the so obtained compounds, in a definite stereochemical relationship, offers the possibility of usefully synthetic manipulations directed towards the synthesis of natural alkaloids.

EXPERIMENTAL

Mps were measured on a Kofler apparatus and are uncorrected. Elemental analyses were performed with a Perkin-Elmer elemental analyzer. Infrared spectra were recorded on a Perkin-Elmer 377 instrument. ¹H NMR spectra were measured on a Varian 300 Gemini instrument in CDCl₃ as solvent. Chemical shifts are in ppm (δ) from TMS as internal standard. NOE difference spectra were obtained by substracting alternatively right-off-resonance free induction decays (FIDs) from right-on-resonance-induced FIDs. Reaction mixtures were analyzed by tlc on silica gel GF 254 (Merck) and the spots were detected under uv light (254 nm). Flash chromatography was carried out with Kieselgel 60 (Merck).

Preparation of *E*-enamido esters derivatives 14a-c.

General procedure. A solution (165 mmol) of trans acyl chloride 13 in 150 ml of anhydrous carbon tetrachloride was added dropwise, at 0 °C, to a stirred solution containing 75 mmol of 12a, b and 22.5 ml (165 mmol) of $E_{13}N$ in 150 ml of anhydrous carbon tetrachloride. The reaction mixture was stirred at 0 °C for 30 min and then at 25 °C for 6 h. The mixture was filtered and washed with 50 ml of carbon tetrachloride. The combined filtrates were washed with water, dried with sodium sulfate, filtered, and the solvent was removed under reduced pressure. The residue was subjected to silica flash-chromatography using a methanol/chloroform 3:97 mixture as eluent.

Reaction of 12a with cinnamoyl chloride. First fractions gave (E,E)-3-(N-isopropyl-N-cinnamoylamino)-propyl cinnamoate 14a. Oil (95%); two rotamers; ir (neat) 3060, 2935, 1750, 1645, 1450, 1375, 1250, 1100, 980, 860 cm⁻¹. ¹H Nmr: δ (CDCl₃) 1.22 (d, 3H, CH₃, J = 6.3 Hz), 1.28 (d, 3H, CH₃, J = 6.6 Hz), 2.09 (m, 2H, CH₂), 3.46 (m, 2H, N-CH₂), 4.29 (m, 2H, OCH₂), 4.87 (ept, 1H, CH(CH₃)₂, J = 6.3 and 6.6 Hz), 6.44, 6.47 (d, total 1H, =CH, J = 15.9 Hz), 6.89, 6.92 (d, total 1H, =CH, J = 15.9 Hz), 7.26-7.53 (m, 10H, ArH), 7.61-7.69 (m, total 2H, =CHPh). ¹³C Nmr: δ (CDCl₃) 22.35, 25.50, 30.45, 42.35, 47.30, 64.50, 118.15, 119.10, 127.70, 127.85, 128.10, 130.50, 130.85, 132.25, 133.50, 135.64, 143.77, 146.92, 168.95, 170.92. MS: m/e (M[†]) 377. (Found: C, 76.12; H, 7.38; N, 3.69%. Calc. for C₂₄H₂₇NO₃: C, 76.35; H, 7.22; N, 3.71%).

Reaction of 12b with cinnamoyl chloride. First fractions gave (E,E)-3-(N-benzyl-N-cinnamoylamino)-propyl cinnamoate 14b. Oil (95%); two rotamers; ir (neat) 3040, 2980, 1740, 1640, 1600, 1550, 1180, 970, 860, 760 cm⁻¹. ¹H Nmr: δ (CDCl₃) 2.05 (m, 2H, CH₂), 3.55 (m, 2H, N-CH₂), 4.35 (m, 2H, OCH₂), 4.76 (s, 2H, PhCH₂), 6.42, 6.45 (d, total 1H, =CH, J = 15.6 Hz), 6.85, 6.89 (d, total 1H, =CH, J = 15.6 Hz), 7.20-7.55 (m, 10H, ArH), 7.78 (m, total 2H, =CHPh). ¹³C Nmr: δ (CDCl₃) 47.53, 50.50, 64.30, 64.54, 116.86, 117.37,

127.55, 127.91, 128.56, 129.33, 130.20, 134.10, 142.02, 142.70, 145.13, 166.38, 166.80. MS: m/e (M^{+}) 425. (Found: C, 79.10; H, 6.41; N, 3.29%. Calc. for $C_{28}H_{27}NO_3$: C, 79.02; H, 6.40; N, 3.29%).

Reaction of 12a with crotonyl chloride. First fractions gave (E,E)-3-(N-isopropyl-N-but-2-enoylamino)-propyl but-2-enoate 14c. Oil (98%); two rotamers; ir (neat) 3350, 2940, 1750, 1665, 1620, 1440, 1375, 1240, 1170, 1095, 970,750 cm⁻¹. ¹H Nmr: δ (CDCl₃) 1.20 (d, 3H, CH₃, J = 6,3 Hz), 1.26 (d, 3H, CH₃, J = 6.6 Hz), 1.68, 1.85 (d, total 6H, CH₃, J = 6.8 Hz), 2.15 (m, 2H, CH₂), 3.40 (m, 2H, N-CH₂), 4.45 (m, 2H, OCH₂), 4.86 (ept, 1H, CH, J = 6.3 and 6.6 Hz), 5.71,5.92 (m, total 2H, =CH), 6.55, 6.85 (m, total 2H, =CH). ¹³C Nmr: δ (CDCl₃) 22.20, 24.35, 26.89, 45.50, 47.45, 50.01, 118.85, 120.05, 139.50, 140.30, 170.70, 171.20. MS: m/e (M¹) 2.53. (Found: C, 65.98; H, 9.15; N, 5.54%. Calc. for C₁₄H₂₃NO₃: C, 66.36; H, 9.16; N, 5.53%).

Preparation of (E)-enamido-propanol derivatives 15a-c.

General procedure. To a stirred solution containing 50 mmol of 12a-c in 280 ml of methanol, 6% aqueous K₂CO₃ (150 ml) was added. The mixture was stirred overnight; after removal of the solvent under reduced pressure, the residue was subjected to silica flash-chromatography (MeOH/CHCl₃ 4:96).

Reaction of 14a with K_2CO_3 . First eluted product was (E)-N-isopropyl-N-(3-propanol)cinnamamide 15a. Oil (90%); ir (neat) 3600-3200, 2970, 1660, 1570, 1400, 1375, 1060, 980,760 cm⁻¹. ¹H Nmr: δ (CDCl₃) 1.26 (d, 6H, CH₃, J = 6.6 Hz), 1.71 (m, 2H, CH₂), 3.51 (m, 4H, OCH₂, NCH₂), 4.29 (sept, 1H, CH₂(CH₃)₂, J = 6.6 Hz), 6.88 (d, 1H, =CH, J = 15.1 Hz), 7.32-7.50 (m, 5H, ArH), 7.65 (d, 1H, =CH, J = 15.1 Hz). ¹³C Nmr: δ (CDCl₃) 22.35, 25.50, 30.45, 42.35, 47.30, 64.50, 118.15, 119.10, 127.70, 127.85, 128.10, 130.50, 130.85, 132.25, 133.50, 135.64, 143.77, 146.92, 168.95, 170.92. MS: nl/e (M⁺) 247. (Found: C, 73.12; H, 8.53; N, 5.69%. Calc. for C₁₅H₂₁NO₂: C, 72.83; H, 8.56; N, 5.67%).

Reaction of 14b with K_2CO_3 . First eluted product was (E)-N-benzyl-N-(3-propanol)cinnamamide 15b. Oil (85%); ir (neat) 3600-3200, 3060, 2980, 1675, 1600, 1400, 1300, 1090, 980, 700 cm⁻¹. ¹H Nmr: δ (CDCl₃) 1.73 (m, 2H, CH₂), 3.48 (bs, 2H, NCH₂), 3.58 (bs, 2H, OCH₂), 4.20 (bs, 1H, OH), 4.60 (s, 2H, NCH₂Ph), 6.74 (d, 1H, =CH, J = 15.3 Hz), 7.17-7.37 (m, 10H, ArH), 7.72 (d, 1H, =CH, J = 15.3 Hz). ¹³C Nmr: δ (CDCl₃) 31.75, 42.73, 44.81, 51.87, 117.10, 125.39, 127.82, 127.91, 128.73, 128.79, 128.96, 129.72, 135.13, 137.52, 143.74, 167.50. MS: m/e (M⁺) 295. (Found: C, 77.23; H, 7.18; N, 4.74%). Calc. for C₁₉H₂₁NO₂: C, 77.25; H, 7.17; N, 4.74%).

Reaction of 14c with K_2CO_3 . First eluted product was (E)-N-isopropyl-N-(3-propanol)but-2-enamide 15c. Oil (80%); ir (neat) 3600-3200, 2980, 1670, 1560, 1400, 1375, 1300, 1060, 970, 875 cm⁻¹. ¹H Nmr: δ (CDCl₃) 1.25 (d, 6H, CH₃, J = 6.6 Hz), 1.65 (m, 2H, CH₂), 1.80 (dd, 3H, CH₃, J = 1.6 and 6.5 Hz), 3.50 (m, 4H, NCH₂ and OCH₂), 3.67 (bs, 1H, OH), 4.27 (sept, 1H, CH, J = 6.6 Hz), 6.09 (d, 1H, =CH, J = 15.0 Hz), 6.40 (dq, 1H, =CH, J = 6.5 and 15.0 Hz). ¹³C Nmr: δ (CDCl₃) 22.53, 24.87, 25.50, 30.40, 41.35, 47.20, 64.50, 119.65, 148.70, 165.60. MS: m/e (M¹) 185. (Found: C, 64.57; H, 10.33; N, 7.58%. Calc. for C₁₀H₁₉NO₂: C, 64.82; H, 10.34; N, 7.56%).

Preparation of (E)-enamide-propanal derivatives 16a-c.

General procedure. 8.5 ml (120 mmol) of anhydrous DMSO were added, at -78 °C, to a stirred solution of bis(trichloromethyl)carbonate (20 mmol) in 50 ml of dry dichloromethane at -78 °C. The reaction mixture was stirred for 15 min and then a solution of **15a-c** (8 mmol) in 80 ml of dichloromethane was slowly added at the same temperature. After 15 min of stirring, triethylamine (19.7 ml, 140 mmol) in 100 ml of dichloromethane was added dropwise maintaining the temperature below -70 °C. After the addition, the resulting suspension was

stirred at -78 °C for 5 min and then the acetone-dry bath was removed. The reaction mixture was stirred at rt for 2 h and the solvent was removed under reduced pressure. The obtained residue was extracted with dichloromethane, washed with water, dried with sodium sulfate and silica flash chromatographated (MeOH/CHCL₃ 3:97).

First eluted product was (*E*)-3-*N*-isopropyl-*N*-(3-propanal)cimnamamide 16a. Oil (60%); ir (neat) 3050, 3020, 2980, 1735, 1660, 1600, 1500, 1375, 1130, 980, 760 cm⁻¹. ¹H Nmr: δ (CDCl₃) 1.23 (d, 3H, CH₃, J = 6.3 Hz), 1.28 (d, 3H, CH₃, J = 6.6 Hz), 2.70 (m, 2H, CH₂), 3.48 (m, 2H, NCH₂), 4.20 (sept, 1H, NCH, J = 6.3 and 6.6 Hz), 6.60 (d, 1H, =CH, J = 16.2 Hz), 7.22-7.58 (m, 2H, ArH and =CH), 9.76 (s, 1H, CHO). ¹³C Nmr: δ (CDCl₃) 22.30, 24.45, 36.34, 59.10, 64.30, 120.51, 127.53, 128.64, 129.49, 134.64, 140.75, 166.96, 201.51. MS: m/e (M⁺) 245. (Found: C, 72.68; H, 7.78; N, 5.70%. Calc. for C₁₅H₁₉NO₂: C, 73.43; H, 7.81; N, 5.71%).

First eluted product was (*E*)-3-*N*-benzyl-*N*-(3-propanal)cinnamamide 16b. Oil (55%); ir (neat) 3060, 3000, 1730, 1660, 1400, 1130, 990, 770 cm⁻¹. ¹H Nmr: δ (CDCl₃) 2.77 (m, 2H, CH₂), 3.50 (m, 2H, NCH₂), 5.01 (s, 2H, NCH₂Ph), 6.59 (d, 1H, =CH, J = 15.1 Hz), 7.20 (m, 6H, ArH and =CH), 9.76 (s, 1H, CHO). ¹³C Nmr: δ (CDCl₃) 36.31, 43.64, 59.15, 123.91, 127.62, 127.78, 128.53, 128.84, 129.10, 129.76, 134.60, 135.39, 139.23, 201.50. MS: m/e (M⁺) 293 (Found: C, 77.24; H, 6.50; N, 4.77%. Calc. for C₁₉H₁₉NO₂: C, 77.78; H, 6.53; N, 4.78%).

First eluted product was (*E*)-3-*N*-isopropil-*N*-(3-propanal) but-2-enamide 16c. Oil (60%); ir (neat) 3050, 3010, 2980, 1730, 1670, 1500, 1375, 1130, 970, 830 cm⁻¹. ¹H Nmr: δ (CDCl₃) 1.24 (d, 3H, CH₃, J = 6.3 Hz), 1.28 (d, 3H, CH₃, J = 6.6 Hz), 1.80 (d, 3H, CH₃, J = 6.8 Hz), 2.68 (m, 2H, CH₂), 3.48 (m, 2H, CH₂), 4.25 (sept, 1H, CH, J = 6.3 and 6.6 Hz), 6.26 (d, 1H, =CH, J = 15.1 Hz), 6.88 (m, 1H, =CH), 9.75 (s, 1H, CHO). ¹³C Nmr: δ (CDCl₃) 22.40, 24.35, 25.60, 36.45, 59.60, 64.80, 118.96, 137.27, 168.19, 206.34. MS: m/e (M⁺) 183 (Found: C, 65.70; H, 9.20; N, 7.66%. Calc. for C₁₀H₁₇NO₂: C, 65.51; H, 9.35; N, 7.64%).

Preparation of 8-oxa-3,7-diazabicyclo[4.3.0]nonan-2-ones 18a-c.

General procedure. A mixture containing 7.5 mmol of compound 16a-c, 11.5 ml (8.25 mmol) of triethylamine, 8.25 mmol of N-substituted hydroxylamines in 200 ml of absolute ethanol was refluxed for 36 h. At the end of this time the solvent was evaporated under reduce pressure and the residue subjected to silica flash-chromatography (MeOH/CHCl₃ 2:98).

Reaction of 16a with N-phenyl hydroxylamine. First fractions gave 7,9-diphenyl-3-isopropyl-8-oxa-3,7-diazabicyclo[4.3.0]nonan-2-one 18a. Oil (65%); ir (neat) 3050, 3010, 2980, 1690, 1600, 1375, 1130, 980, 750 cm⁻¹. ¹H Nmr: δ (CDCl₃) 1.07 (d, 3H, CH₃, J = 6.9 Hz), 1.14 (d, 3H, CH₃, J = 6.6 Hz), 1.95 (m, 2H, 5-CH₂), 3.13 (m, 1H, H₄), 3.41 (dd, 1H, H₁, J = 6.4 and 9.0 Hz), 3.42 (m, 1H, H₄), 3.82 (m, 1H, H₆), 4.86 (sept, 1H, CH, J = 6.6 and 6.9 Hz), 5.17 (d, 1H, H₉, J = 6.4 Hz), 7.13, 7.53 (m, 10H, ArH). ¹³C Nmr: δ (CDCl₃) 19.29, 19.51, 36.46, 44.07, 56.41, 63.28, 81.88, 116.51, 118.24, 123.67, 126.65, 127.89, 128.40, 128.39, 128.92, 139.74, 148.10, 167.89. MS: m/e (M⁺) 336. (Found: C, 74.92; H, 7.12; N, 8.31%. Calc. for C₂₁H₂₄N₂O₂: C, 74.96; H, 7.19; N, 8.33%).

Reaction of **16b** with N-methyl hydroxylamine. First fractions gave 3-benzyl-7-methyl-9-phenyl-8-oxa-3,7-diazabicyclo[4.3.0]nonan-2-one **18b**. Oil (65%); ir (neat) 3010, 2970, 1680, 1610, 1350, 1130, 980, 760 cm⁻¹.

¹H Nmr: δ (CDCl₃) 1.76 (m, 1H, H₃), 1.94 (m, 1H, H₃), 2.75 (s, 3H, NCH₃), 3.06 (m, 2H, H₄ and H₆), 3.28 (dd, 1H, H₁, J = 6.4 and 8.4 Hz), 3.57 (ddd, 1H, H₄, J = 3.3, 8.1 and 12.0 Hz), 4.51 (d, 1H, CH₂Ph, J = 14.7 Hz), 4.73 (d, 1H, CH₂Ph, J = 14.7 Hz), 5.02 (d, 1H, H₉, J = 6.4 Hz), 7.23-7.56 (m, 10H, ArH).

¹³C Nmr: δ (CDCl₃) 24.38, 42.55, 43.10, 50.35, 56.30, 66.16, 82.19, 126.73, 127.49, 127.94, 128.00, 128.39, 128.65, 136.68,

169.15. MS: m/e (M $^{+}$) 322. (Found: C, 74.40; H, 6.85; N, 8.70%. Calc. for $C_{20}H_{22}N_2O_2$: C, 74.50; H,6.88; N, 8.69%).

Reaction of 16c with N-methyl hydroxylamine. First fractions gave 7,9-dimethyl-3-isopropyl-8-oxa-3,7-diazabicyclo[4.3.0]nonan-2-one 18c. Oil (70%); ir (neat) 3020, 3000, 2980, 1680, 1600, 1375, 1130, 980 cm⁻¹. ¹H Nmr: δ (CDCl₃) 1.07 (d, 3H, CH₃, J = 4.5 Hz), 1.10 (d, 3H, CH₃, J = 4.5 Hz) 1.45 (d, 3H, CH₃, J = 6.0 Hz), 1.78 (m, 2H, CH₂), 2.68 (s, 3H, NCH₃), 2.80 (dd, 1H, H₁, J = 6.9 and 9.3 Hz), 2.88 (m, 1H, H₆), 3.08 (ddd, 1H, H₄, J = 3.9, 4.2 and 12.9 Hz), 3.38 (ddd, 1H, H₄, J = 3.6, 4.0 and 12.9 Hz), 3.80 (dq, 1H, H₉, J = 6.0 and 6.9 Hz), 4.87 (ept, 1H, NCH, J = 4.5 Hz). ¹³C Nmr: δ (CDCl₃) 19.02, 19.39, 25.53, 35.58, 43.03, 43.52, 55.42, 65.65, 77.33, 168.14. MS: m/e (M⁺) 212. (Found: C, 62.12; H, 9.39; N, 13.17%. Calc. for C₁₁H₂₀N₂O₂: C, 62.22; H, 9.50; N, 13.20%).

Preparation of 8-oxa-3,9-diazabicyclo[4.3.0]nonan-4-ones 23b,c and 24a,b.

General procedure. To a stirred solution containing 75 mmol of methylamino acetaldehyde dimethyl acetal 19, and 11.25 of triethylamine in 50 ml of dry carbon tetrachloride was added dropwise a solution of 82.5 mmol of corresponding acyl chloride in 50 ml of dry carbon tetrachloride at 0 °C. The solution was strirred at 25 °C for 6 h and then filtered. The obtained solid was washed with 30 ml of carbon tetrachloride, the combined filtrate washed with 10 ml of water, and dried with sodium sulfate. The solvent was removed at reduced pressure, and the residue was subject to silica flash-chromatography using a methanol/chloroform 2:98 mixture as eluent.

N-Methyl-N-(acetaldehyde dimethyl acetal)but-3-enamide **20a**. Oil (70%); ir (neat) 2940, 2840, 1690, 1460, 1410, 1280, 1190, 1105, 1075, 980, 800, 725 cm⁻¹. ¹H Nmr: δ (CDCl₃) 2.90, 3.00 (s, total 3H, N-CH₃), 3.20 (m, 2H, CH₂CO), 3.40, (s, 6H, OCH₃), 3.45 (d, 2H, NCH₂, J = 5.3 Hz), 4.35, 4.45 (t, total 1H, OCH, J = 5.3 Hz), 5.18 (m, 2H, =CH₂), 5.80 (m, 1H, =CH).). ¹³C Nmr: δ (CDCl₃) 39.45, 39.51, 53.51, 55.86, 104.40, 118.89, 132.34, 172.30. MS: m/e (M⁺) 187. (Found: C, 58.25; H, 9.17; N, 7.46%. Calc. for C₉H₁₇NO₃: C, 57.72; H, 9.16; N, 7.48%).

(E)-N-methyl-N-(acetaldehyde dimethyl acetal)pent-3-enamide 20b. Oil (80%); ir (neat) 2965, 2830, 1680, 1630, 1410, 1150, 1100, 960 cm⁻¹. ¹H Nmr: δ (CDCl₃) 1.68 (dd, 3H, CH₃, J = 6.3 and 1.8 Hz), 2.96, 3.05 (s, total 3H, NCH₃), 3.18 (m, 2H, OCH₂), 3.39 (s, 6H, OCH₃), 3.42 (d, 2H, NCH₂, J = 5.1 Hz), 4.40, 4.50 (t, total 1H, OCH, J = 5.1 Hz), 5.55 (m, 2H, =CH). ¹³C Nmr: δ (CDCl₃) 17.80, 37.10, 39.80, 54.40, 58.40, 105.30, 122.70, 137.00, 173.10. MS: m/e (M⁺) 201. (Found: C, 59.60; H, 9.51; N, 6.96%. Calc. for C₁₀H₁₉NO₃: C, 59.66; H, 9.52; N, 6.96%).

(E)-N-methyl-N-(acetaldehyde dimethyl acetal)-4-phenylbut-3-enamide 20c. Oil (90%); ir (neat) 2975, 2950, 1670, 1605, 1410, 1200, 1040, 980, 760 cm⁻¹. ¹H Nmr: δ (CDCl₃) 2.93, 3.03 (s, total 3H, NCH₃), 3.23 (d, 2H, CH₂CO, J = 6.6 Hz), 3.30, (s, 6H, OCH₃), 3.40 (d, 2H, NCH₂, J = 5.7 Hz), 4.37, 4.45 (t, total 1H, OCH), 6.28, 6.43 (m, 2H, =CH), 7.15-7.31 (m, 5H, ArH). ¹³C Nmr: δ (CDCl₃) 38.37, 38.70, 50.94, 53.21, 55.21, 55.91, 103.55, 123.54, 124.19, 126.86, 128.05, 129.05, 133.41, 137.63, 167.40. MS: m/e (M⁺) 263. (Found: C, 68.51; H, 8.10; N, 5.46%. Calc. for C₁₃H₂₁NO₃: C, 68.39; H, 8.04; N, 5.31%).

N-Methyl-N-(acetaldehyde)but-3-enamide **21a**. Oil (40%); ir (neat) 2940, 2860, 1730, 1680, 1580, 1190, 1140, 1080, 985 cm⁻¹. ¹H Nmr: δ (CDCl₃), 2.98, 3.07 (s, total 3H, NCH₃), 3.21 (d, 2H, CH₂, J = 5.4 Hz) 4.18 (s, 2H, NCH₂), 5.15 (m, 2H, =CH₂), 5.89 (m, 1H, =CH), 9.55 (s, 1H, CHO). ¹³C Nmr: δ (CDCl₃) 38.87, 54.41, 58.70, 118.70, 131.29, 168.30, 197.76. MS: m/e (M⁺) 141. (Found: C, 58.12; H, 7.91; N, 9.94%. Calc. for C₇H₁₁NO₂: C, 59.54; H, 7.86; N, 9.93%).

(E)-N-methyl-N-(acetaldehyde)pent-3-enamide 21b. Oil (70%); ir (neat) 2930, 2870, 2730, 1735, 1670,

1405, 1295, 1230, 1110, 980, 830 cm⁻¹. ¹H Nmr: δ (CDCl₃) 1.70 (dd, 3H, CH₃, J = 1.8 and 6.3 Hz), 3.06 (s, 3H, NCH₃), 3.13 (m, 2H, CH₂), 4.17 (s, 2H, NCH₂), 5.54 (m, 2H, =CH), 9.54 (s, 1H, CHO). ¹³C Nmr: δ (CDCl₃) 18.48, 37.75, 54.41, 58.60, 123.70, 137.05, 173.10, 197.98. MS: m/e (M⁺) 155. (Found: C, 61.92; H, 8.40; N, 9.10 %. Calc. for C₈H₁₃NO₂: C, 61.89; H, 8.45; N, 9.03 %).

(E)-N-Methyl-N-(acetaldehyde)-4-phenyl-but-3-enamide 21c. Oil (40%); ir (neat) 3060, 2980, 2950, 1740, 1650, 1400, 1200, 980, 760, 680 cm⁻¹. ¹H Nmr: δ (CDCl₃) 3.05 (s, 3H, NCH₃), 3.40 (d, 2H, CH₂, J = 5.4 Hz), 4.18 (s, 2H, NCH₂), 6.32 (m, 2H, =CH), 7.30-7.56 (m, 5H, ArH), 9.50 (s, 1H, CHO). ¹³C Nmr: δ (CDCl₃) 38.70, 56.20, 58.30, 125.30, 128.76, 129.35, 133.41, 137.63, 168.50, 198.30. MS: m/e (M⁺) 217. (Found: C, 71.80; H, 6.97; N, 6.45%. Calc. for C₁₃H₁₅NO₂: C, 71.85; H, 6.96; N, 6.45%).

3,7,9-Trimethyl-8-oxa-3,9-diazabicyclo[4.3.0]nonan-4-one 23b. Oil (40%); ir (neat) 2980, 2920, 1670, 1480, 1280, 1060 cm⁻¹. ¹H Nmr: δ (CDCl₃) 1.29 (d, 3H, CH₃, J = 6.1 Hz), 2.28 (dd, 1H, H₅, J = 3.6 and 15.3 Hz), 2.41 (dd, 1H, H₅, J = 6.3 and 15.3 Hz), 2.47 (m, 1H, H₆), 2.77 (s, 3H, NCH₃), 3.01 (m, 4H, NCH₃ and H₁), 3.13 (dd, 1H, H₂, J = 2.7 and 13.8 Hz), 3.48 (dd, 1H, H₂, J = 3.6 and 13.8 Hz), 3.69 (dq, 1H, H₇, J = 4.0 and 6.1 Hz). ¹³C Nmr: δ (CDCl₃) 17.55, 33.37, 36.10, 44.65, 48.58, 48.60, 50.30, 68.63, 78.69, 171.37. MS: m/e (M¹) 184. (Found: C, 58.61; H, 8.79; N, 15.18%. Calc. for C₉H₁₆N₂O₂: C, 58.66; H, 8.76; N, 15.21%).

3,9-Dimethyl-8-oxa-7-phenyl-3,9-diazabicyclo[4.3.0]nonan-4-one 23c. Oil (70%); ir (neat) 3080, 2920, 2900, 1670, 1480, 1440, 1290, 1050, 750, 700 cm⁻¹. 1 H Nmr: δ (CDCl₃) 2.41 (m, 2H, H₅), 2.87 (m, 4H, NCH₃ and H₆), 3.05 (s, 3H, CH₃), 3.25 (m, 2H, H₁ and H₂), 3.54 (dd, 1H, H₂, J = 3.6 and 14.8 Hz), 4.52 (d, 1H, H₇, J = 8.7 Hz), 7.21-7.41 (m, 5H, ArH). 13 C Nmr: δ (CDCl₃) 32.85, 36.70, 44.50, 49.46, 50.12, 68.54, 84.82, 126.81, 127.44, 129.14, 137.69, 173.10. MS: m/e (M⁺) 246. (Found: C, 67.89; H, 7.39; N, 11.38%. Calc. for C₁₄H₁₈N₂O₂: C, 68.25; H, 7.37; N, 11.38%).

3,9-Dimethyl-8-oxa-3,9-diazabicyclo[4.3.0]nonan-4-one 24a. Oil (60%); ir (neat) 2970, 2960, 1660, 1470, 1270, 920 cm⁻¹. ¹H Nmr: δ (CDCl₃) 2.26 (dd, 1H, H₅, J = 4.5 and 15.0 Hz), 2.37 (dd, 1H, H₅·, J = 5.7 and 15.0 Hz), 2.65 (s, 3H, NCH₃), 2.90 (s, 3H, NCH₃), 3.00 (m, 2H, H₁ and H₆), 3.14 (dd, 1H, H₂, J = 3.9 and 13.8 Hz), 3.41 (m, 1H, H₂), 3.44 (m, 1H, H₇), 4.07 (dd, 1H, H₇·, J = 7.5 and 8.4 Hz). ¹³C Nmr: δ (CDCl₃) 20.10, 34.88, 36.16, 41.72, 44.95, 51.00, 67.58, 71.91, 171.71. MS: m/e (M⁺) 170. (Found: C, 55.67; H, 8.21; N, 16.50%. Calc. for C₈H₁₄N₂O₂: C, 56.44; H, 8.29; N, 16.46%).

3,7,9-Trimethyl-8-oxa-3,9-diazabicyclo[4.3.0]nonan-4-one 24b. Oil (10%); ir (neat) 2980, 2910, 1675, 1470, 1280, 1060 cm⁻¹. ¹H Nmr: δ (CDCl₃) 1.18 (d, 3H, CH₃, J = 6.1 Hz), 2.23 (dd, 1H, H₅, J = 9.3 and 15.0 Hz) 2.39 (dd, 1H, H₅, J = 9.6 and 15.0 Hz), 2.71 (s, 3H, NCH₃), 2.95 (m, 5H, NCH₃, H₁ and H₆), 3.34 (m, 2H, H₂), 4.44 (dq, 1H, H₇, J = 6.1 and 6.4 Hz). ¹³C Nmr: δ (CDCl₃) 14.67, 23.45, 32.00, 37.80, 44.64, 52.24, 68.02, 72.10, 74.70, 172.62. MS: m/e (M⁺) 184. (Found: C, 59.99; H, 8.70; N, 15.16%. Calc. for C₉H₁₆N₂O₂: C, 58.66; H, 8.76; N, 15.21%).

Preparation of piperidines derivatives.

The above compounds were prepared according to the general method already reported by us.

4-(N-Methylamino)-2-dideutero-3-hydroxybenzyl-N-benzylpiperidine 25b. Oil (60%); ir (neat) 3420, 3295, 3010, 2960, 1430, 1370, 1130, 1050, 980, 750 cm⁻¹. ¹H Nmr: δ (CDCl₃) 1.81-1.98 (m, 2H, H₅), 2.39-2.41 (m, 1H, H₆), 2.54, 2.67 (m, 2H, H₆ and H₃), 2.82 (s, 3H, NCH₃), 2.85-2.93 (m, 1H, H₄), 3.47 (d, 1H, CH₂Ph, J = 13.2 Hz), 3.58 (d, 1H, CH₂Ph, J = 13.2 Hz), 4.91(m, 1H, CHOH), 7.20-7.34 (m, 10H, ArH). ¹³C Nmr: δ (CDCl₃) 27.51, 29.67, 45.19, 49.56, 50.27, 62.82, 64.27, 82.50, 127.04, 127.44, 128.08, 128.21, 128.31, 128.52, 128.62, 128.93, 138.41. MS: m/e (M⁺) 312. (Found: C, 78.65; H, 7.59; D, 1.30; N, 9.00%. Calc. for

C₂₀H₂₄D₂N₂O: C, 76.88; H, 7.74; D, 1.29; N, 8.97%).

4-(N-Methylamino)-2-dideutero-3-(1-ethanol)-N-isopropylpiperidine 25c. Oil (50%); ir (neat) 3600-3290, 1450, 1140, 1050 cm⁻¹. ¹H Nmr: δ (DMSO d₆) 0.93 (d, 6H, CH₃, J = 6.6 Hz), 1,15 (d, 3H, CH₃, J = 6.3 Hz), 1.56 (m, 1H, H₅), 1,70 (m, 1H, H₅·), 2.20 (m, 1H, H₆), 2.30 (m, 2H, H₃ and H₆·), 2.64 (s, 3H, NCH₃), 2.66 (m, 2H, H₄ and CHⁱ), 3.33 (bs, 2H, OH and NH), 3.73 (m, 1H, CHOH). ¹³C Nmr: δ (DMSO d₆) 17.88, 21.72, 26.03, 29.05, 30.46, 34.39, 47.81, 53.72, 63.55, 75.60. MS: m/e (Mⁱ) 202. (Found: C, 65.21; H, 10.92; D, 1.98; N, 13.82%. Calc. for C₁₁H₂₂D₂N₂O: C, 65.30; H, 10.96; D, 1.99; N, 13.85%).

3-(N-Methylamino)-6-dideutero-4-(1-ethanol)-N-methylpiperidine 26. Oil (40%); ir (neat) 3500-3200, 1470, 1150, 1060 cm⁻¹. ¹H Nmr: δ (DMSO-d₆) 1.20 (d, 3H, CH₃, J = 6.0 Hz), 1.71 (dd, 1H, H₅, J = 7.6 and 13.7 Hz), 1.89 (dd, 1H, H₅, J = 6.4 and 13.7 Hz), 2.28 (m, 2H, H₃ and H₄), 2.60 (s, 6H, NCH₃), 2.94, 2.98 (m, 2H, H₂), 3.82 (dq, 1H, CHOH, J = 5.1 and 6.0 Hz). ¹³C Nmr: δ (DMSO-d₆) 21.27, 23.36, 29.02, 31.05, 43.27, 44.02, 47.66, 63.20, 76.89. MS: m/e (M⁻¹) 174. (Found: C, 61.78; H, 10.49; D, 2.30; N, 16.09%. Calc. for C₉H₁₈D₂N₂O: C, 62.03; H, 10.41; D, 2.31; N, 16.07%).

ACKNOWLEDGEMENTS

Authors are grateful to the Italian M. U. R. S. T. and C. N. R. for partial financial support.

REFERENCES AND NOTES

- Bernet, B.; Vasella, A. Helv. Chim. Acta 1979, 62, 1990-2016. Tufariello, J. J. Acc. Chem. Res. 1979, 12, 396-403.. Carruthers, V. Cycloaddiction Reactions in Organic Synthesis, Baldwin, J. E. Ed.; FRS & P. D. Magnus, FRS, Tetrahedron Organic Chemistry Series, Vol. 8, 1990; pp. 269-331. Ihara, M.; Takahashi, M.; Fukumoto, K.; Kametani, T. J. Chem Soc., Chem. Commun. 1988, 9, 10. McCaig, A. E.; Wightman, R. H. Tetrahedron Lett., 1993, 34, 3939-3942. Confalone, P. N.; Huie, E. M. Org. React. 1988, 36, 1-173. Padwa, A.; Schoffstall, A. M. "Advances in Cycloaddition", Vol. 2, ed. by D. P. Curran, JAI Press Inc., Greenwich, 1990, pp. 2-28. Hassner, A. Heterocycles in Bio-Organic Chemistry; Pergmon, J. Ed.; Royal Soc. of Chem.: Cambridge, 1991; pp. 130-143. Grünanger, P.; Vita-Finzi, P. The Chemistry of Heterocyclic Compounds; Taylor, E.C., Ed.; Wiley-Interscience: New York, 1991, vol 49, pp. 753-774.
- Hassner, A.; Murthy, K. S. K.; Padwa, A.; Chiacchio, U.; Dean, D. C.; Schoffstal, A. M. J. Org. Chem. 1989, 54, 5277. Buemi, G.; Chiacchio, U.; Corsaro, A.; Rescifina, A.; Romeo, G.; Uccella, N.; Hassner, A. Heterocycles 1993, 36, 2005. Chiacchio, U.; Casuscelli, F.; Corsaro, A.; Librando, V.; Rescifina, A.; Romeo, R.; Romeo, G. Tetrahedron 1995, 51, 5689. Chiacchio, U.; Corsaro, A.; Pistarà, V.; Rescifina, A.; Romeo, G.; Romeo, R. Tetrahedron 1996, 52, 7875.
- LeBel, N. A.; Whang, J. J. Am. Chem. Soc. 1959, 81, 6334-6335. LeBel, N. A. Ann. N. Y. Acad. Sci. 1965, 27, 858.
- Baldwin, S. W.; Wilson, J. D.; Aubé, J. J. Org. Chem. 1985, 50, 4432. Baldwin, S. W.; McFadyen, R. B.; Aubé, J.; Wilson, J. D. Tetrahedron Lett. 1991, 32, 4431-4434. Baldwin, S. W.; Gedon, S. C. Synt. Comm. 1991, 21, 587-596.
- 5. LeBel, N. A.; Banucci, E. G. J. Org. Chem. 1971, 36, 2440.
- 6. Chiacchio, U.; Buemi, G.; Casuscelli, F.; Procopio, A.; Rescifina, A.; Romeo, R. *Tetrahedron* 1994, 50, 5503.
- 7. Elbein, A. D.; Molyneux, R. J. In Alkaloids: Chemical and Biological Perspectives; Pelletier, S. W. Ed.;

Wiley: New York, 1987; Vol. 5, pp. 1-54. Casiraghi, G.; Zanardi, F.; Rassu, G.; Spanu, P. Chem. Rev. 1995, 95, 1677.

8. Elderfield, R. C.; Gensler, W. J.; Brady, F.; Head, J. D. J. Am. Chem. Soc. 1946, 68, 1582.

(Received in UK 12 August 1996; accepted 26 September 1996)